Postprint of: Sulej A.M., Polkowska Ż., Astel A., Namieśnik J., Analytical procedures for the determination of fuel combustion products, anti-corrosive compounds, and de-icing compounds in airport runoff water samples, Talanta, Vol. 117 (2013), pp. 158-167, DOI: 10.1016/j.talanta.2013.07.045

© 2013. This manuscript version is made available under the CC-BY-NC-ND 4.0 license https://creativecommons.org/licenses/by-nc-nd/4.0/

Analytical procedures for the determination of fuel combustion products, anti-corrosive

³ compounds, and de-icing compounds in airport runoff water samples

- Anna Sulej^{*}, Żaneta Polkowska^{*}, Aleksander Astel^{**}, Jacek Namieśnik^{*}
- ^{*}Gdańsk University of Technology, Chemical Faculty, Department of Analytical Chemistry, 11/12 G. Narutowicza
- 7 Str., 80- 233 Gdańsk, Poland, phone: +48 58 347 1010, fax: +48 58 347 2694
- **Pomeranian Academy, Biology and Environmental Protection Institute, Environmental Chemistry Research Unit, 8
- 22a Arciszewskiego Str., 76-200 Słupsk, Poland

Author to whom correspondence should be addressed; e-mail: anna.maria.sulej@gmail.com;

10 11 Tel.: +48-58-347-1010; Fax: +48-58-347-2694.

¹² Abstract

13

9

4

The purpose of this study is to propose and evaluate new procedures for determination of fuel 14 combustion products, anti-corrosive and de-icing compounds in runoff water samples collected 15 from the airports located in different regions and characterized by different level of the activity 16 expressed by the number of flights and the number of passengers (per year). The most difficult 17 step in the analytical procedure used for the determination of PAHs, benzotriazoles and glycols is 18 sample preparation stage, due to diverse matrix composition, the possibility of interference 19 associated with the presence of components with similar physicochemical properties. In this 20 study, five different versions of sample preparation using extraction techniques, such as: LLE and 21 SPE, were tested. In all examined runoff water samples collected from the airports, the presence 22 of PAH compounds and glycols was observed. In the majority of the samples, BT compounds 23 were determined. Runoff water samples collected from the areas of Polish and British 24 international airports as well as local airports had similar qualitative composition, but quantitative composition of the analytes was very diverse. New and validated analytical methodologies ensure that the necessary information for assessing the negative impact of airport activities on the environment can be obtained.

Key words: sample preparation techniques, analytical procedures, airport runoff water,
polycyclic aromatic hydrocarbons, benzotriazoles, glycols

29

30 1. Introduction

31 In recent years, there has been a massive expansion within the air transport industry. The air 32 transport industry is likely to continue growing in the long term, which leads to increased levels 33 of waste production [1-4]. One of the most important environmental effects associated with 34 airport operations is the large volume of produced polluted airport runoff water (stormwater 35 runoff) [1-3,5-15]. Runoff water, as an effect of the transformation of atmospheric precipitation, 36 is one of the most important routes through which atmospheric pollutants reach the surface of the 37 Earth. In an airport area the rainfall, which contains atmospheric pollutants, additionally washes 38 over highly polluted surfaces such as: de-icing runways, taxiways, maintenance surface, or runways on the airport platform, and rinses off pollutants from them (Table 1) [3-5, 9-20,16-18]. 39 40 The rainfall waters which turn into runoff waters carry various toxic compounds and get into 41 communal sewage systems, and thence to treatment plants, or, if such plants are non-existent or 42 not working properly, into soil, surface water, and even groundwater which can be the source of 43 drinking water [4,19-26]. A special threat to all elements of the environment are compounds from 44 the glycol group, the benzotriazole group (BTs), and the group of polycyclic aromatic 45 hydrocarbons (PAHs), as they are characterized by high toxicity and cancerogenicity 46 [2,5,8,11,16,19,24,27-31]. Aircraft de-icing and anti-icing fluids are used heavily worldwide, 47 with millions of litres of ADAFs entering the environment every year. The Environmental Protection Agency (U.S EPA) has estimated that approximately 80 million litres of ethylene or propylene glycol-based ADAF-contaminated runoff is discharged directly to surface waters in the United States annually [5,22]. Compounds from the group of benzotriazoles are commonly added to aircraft de-icing fluids as corrosion inhibitors. e.g., in engine coolants, aircraft de-icers, or antifreezing liquids [2,6,7,32].

The annual usage of BTs has been estimated to be about 9000 tons/year in the United States, and global usage is much greater [11,33,34]. Benzotriazoles is the fourth most abundant individual aquatic contaminant (after ethylenediaminetetraacetic acid (EDTA), nitrilotriacetic acid (NTA), and linear alkylbenzene sulfonates (LAS) [25]. According to Directive 67/548/EWG, those compounds are classified as dangerous to the environment and can cause long-term adverse effects in the aquatic environment [4,6,10,22,27-29,33,34].

59 In view of this, it is important to develop new analytical procedures for determination the 60 most important and probably also the most toxic compounds in samples of airport runoff water 61 and to apply the obtained data to assessment of the threats the contaminants pose to surface water 62 and groundwater [1,2,20,27,35-38]. There is no doubt that the most crucial step of suitable 63 analytical protocols is sample preparation for determination of trace and ultratrace constituents. 64 The preparation of samples of airport runoff water for analysis is not a simple task because of: the 65 diversity of compounds in it (analytes), the diversified content of the matrix of the samples, the 66 possibility of interferences related to the occurrence of compounds, which have similar physical 67 and chemical characteristics, in water, and the lack of references necessary to ensure an 68 appropriate quality control/quality assurance (QA/QC) [39]. Only few data have been published 69 on the results of the sample preparation step in runoff water analysis. In the world literature the 70 solid phase extraction (SPE) and liquid-liquid extraction (LLE) techniques were mainly applied 71 to the determination of target analytes in urban runoff water samples [7,9,12,40,41], and the same 72 two extraction techniques were applied in the sample preparation step in airport runoff water 73 analysis [22,23,29,42]. There have also been some reports on the determination of PAHs, BTs 74 and glycols in properly prepared samples with use of gas chromatography (GC), mass 75 spectrometry (MS), chromatography, tandem mass spectrometry (GC-MS/MS), liquid 76 chromatography (LC)-MS and LC-MS/MS, gas chromatography with flame-ionization detection 77 (GC-FID), and two-dimensional gas chromatography coupled to time-of-flight mass spectrometry 78 (GCxGC-TOF-MS) [4,6,16,22,25,27]. The first pieces of information about runoff water 79 analytics have appeared in literature but the problem is still far from being recognized and 80 popularized [22].

81 The purpose of this study is to propose and evaluate new procedures for determination of 82 trace amounts of wide spectrum of xenobiotics in runoff water samples collected from the 83 airports located in different regions and characterized by different level of the activity expressed 84 by the number of flights and the number of passengers (per year). At the step of isolation and 85 preconcentration of analytes from the above-mentioned samples, different versions of LLE and 86 SPE techniques have been applied. The developed analytical methods were used for the 87 determination of PAH compounds, benzotriazoles and glycols, which are the main pollutants at 88 the airports. It seems to be the first work on such a large scale regarding the wide spectrum of 89 analytes and the diversity of places, from which runoff water samples were collected. 90 Multidimensional data have been explored with use of chemometric techniques (Factor Analysis, 91 FA) in order to gain additional information and find correlations between different analytes and 92 parameters of the analyzed samples.

93

95 **2. Materials and methods**

96 2.1 Chemicals and materials

97 Reagents and apparatus used during development of the analytical procedures for the
98 determination of selected compounds (PAHs, BTs, glycols) are shown in Table 2.

99 2.2 Sample collection

100 Runoff water samples were collected during or shortly after the rainfall from the areas of three 101 airports (international airport in Poland, local airport in Poland, and international airport in the 102 United Kingdom). The samples were collected from the areas of the airports in three seasons: 103 autumn, winter, and spring, from 2011 to 2012. During this period 189 runoff water samples 104 were collected from places in which runoff water was lying on lower ground and from airport 105 drainage areas. The places of sample collection were located where the most maintenance work 106 was carried out: fuelling, loading and unloading (of transport airplanes), spraying the airplanes 107 with de-icing substances, parking and servicing of maintenance cars, i.e. the places from which 108 the greatest number of contaminants gets into the runoff waters which later flow into drainage 109 ditches and further into the environment (Table 3). Airport runoff water samples were collected 110 in 1,000 mL bottles of dark glass using a syringe (100 ml) with Teflon tubes. The samples were 111 transported to the laboratory (usually within 1 h after collection). Prior to use, the syringes and 112 tubing were rinsed with MilliQ water and then with the water to be sampled. The runoff samples 113 were usually contaminated with solids (sand, leaves, *etc.*) which had to be pre-filtered (0.45 μ m, 114 Millex®-HV). Bottles were stored at 4 °C in the dark until extraction [43-47].

115 2.3 Development of analytical procedures

116 The desire to obtain reliable information about the state of individual elements of the 117 environment and the processes that occur in them very often requires the use of complex, labourand time-consuming analytical procedures. Therefore, there is a constant need to develop the various stages of the procedures, which will allow to obtain data on the content of trace and ultratrace components in the samples characterized by complex matrices (such as runoff waters from the area of the airports).

122 2.3.1 Sample preparation

123

In the case of airport runoff water samples (where there are a variety of processes associated with the ongoing maintenance of the airport) the sample preparation step is crucial in the analytical procedure. The collected samples should be transported to the laboratory and analyzed as soon as possible (interactions between components in the sample). The decision on the way of sample preparation depends on the type of the obtained analytical information. The first step used in the preparation of environmental samples was to remove solid contaminants (e.g., sand, leaves, dust, etc.) by filtration (pore size 0.45 µm).

Due to the complex composition of the matrix and a large number of interfering compounds, selecting the appropriate extraction method can contribute to optimal (required for further steps in the procedure) sample enrichment, removal of the interfering components and shorter exploitation of apparatus used to perform the final determination. The extraction step is important in view of the fact that errors made at this stage may significantly affect the results of the final determination.

MOST WIEDZY Downloaded from mostwiedzy.pl

For the study of pollutants released to runoff waters in various processes carried out in the areas of airports (fuel combustion products, anti-corrosive and de-icing substances) three groups of compounds: polycyclic aromatic hydrocarbons, benzotriazoles and glycols, were selected. In order to prepare the samples for the determination of organic pollutants, different extraction techniques: liquid-liquid extraction (A1), and solid phase extraction (A2, A3, A4, A5) were

142	applied. Schematic presentation of the analytical procedures used for the determination of
143	compounds from the group of PAHs, BTs and glycols is shown in Figure 1. The recovery and
144	precision for the whole analytical procedure were evaluated by analyzing extracts of runoff water
145	by GC-MS, spiked before extraction with 20 μ gL-1 to 50 μ gL-1 of PAH and BT analytes, and 5
146	and 50 mgL-1 of the glycol analytes. Table 4 shows the average recoveries together with the
147	standard deviation (RSD) of the procedures and comparison of analytical protocols used for the
148	determination of PAHs, BTs and glycols. In addition, the results of PAH determination were
149	explored using the technique of factor analysis (algorithm using principal component factors and
150	varimax rotation of factor loadings). Prior to factor analysis the raw data were auto-scaled by
151	variables as it is commonly recommended in case of data with serious differences in
152	concentration values or different units [48]. The purpose of such pre-processing step was to scale
153	variance of each variable to unity and to centre mean values. The aim of the FA was to identify a
154	structure of the data including three different extraction techniques (A1, A2, A3). The structure of
155	the data was explored by 4 factors that cumulatively explain nearly 80% of the variance (Table
156	5). The first factor indicate a correlation between the content of fluorene, pyrene and benzo [b]
157	fluoranthene, second one between naphthalene, acenaphthene, anthracene, and dibenzo [a, h]
158	anthracene, third one between benzo [a] anthracene, chrysene and fluoranthene and the fourth one
159	between benzo [k] fluoranthene, benzo [a] pyrene, indeno [1,2,3-c, d] terylene and benzo [g, h, i]
160	terylene. The projection of sample distribution in the space of principal components was made
161	with regard to the type of sample preparation technique for the analysis (Figure 2). On the basis
162	of the information shown in Figure 2, which contained the factorial combination of all the
163	factors, it can be concluded that the results of PAH concentrations obtained by means of
164	extraction techniques (A1, A2 and A3) for analytes forming factors (F1, F3 and F4) are
165	comparable. This is evidenced by the general lack of clearly disjoint groups of points

166 corresponding to the three considered extraction techniques. The exception is the score plot 167 shown in Figure 2a, where the results of the analysis of the samples extracted using A3 technique 168 form (at the top of the chart) the group, which is separated from the results of the analysis of the 169 samples prepared by means of A1 and A2 method. It can be concluded that generally higher 170 concentration levels of compounds such as naphthalene, acenaphthene, anthracene, and dibenzo 171 [a, h] anthracene are achieved by application of extraction method A3, compared with the results 172 obtained for the samples extracted by means of A1 and A2 techniques, regardless of the sampling 173 place.

174 2.3.2 Chromatographic analysis of suitably prepared samples

175 Gas chromatography coupled with mass spectrometry (GC-MS) was used for the determination 176 of PAH compounds, glycols and benzotriazoles in the appropriately prepared extracts. The 177 conditions of final determination of the analytes belonging to these groups of compounds were 178 optimized. As a part of this work calibration step, which is aimed to present the dependence 179 between the analytical signal and the concentration of the analyte in a form of a calibration 180 graph, was also performed. On the basis of the calibration curves, it was possible to determine 181 concentration levels of certain substances in real samples. The curves were prepared immediately 182 prior to each series of analysis by diluting the stock solution. Each solution was analyzed in triplicate. Technical specifications, operating conditions and basic metrological parameters of the 183 184 optimized analytical procedures are summarized in Table 6. Calibration was performed in the range of 0.04-2 $[\mu g \cdot L^{-1}]$ for PAHs, 2-300 $[\mu g \cdot L^{-1}]$ for glycols, and 5-75 $[\mu g \cdot L^{-1}]$ for BT, 185 respectively. Linear calibration curves were obtained by plotting the peak area against the 186 187 concentration of the respective standards. Calibrations showed good linearity as indicated by the values of R². The limit of detection (LOD) was determined for glycols, BT, PAHs in quality 188 189 control samples based on three replicates of measurement. LOD was calculated using the equation LOD=3.3SD/b (b is the slope of the calibration curve; SD is the standard deviation of the curve). The quantification limit (LOQ) was set to three times the LOD. The LODs of the PAHs ranged from 0.0003 to $0.057[\mu g \cdot L^{-1}]$, and from 0.0003 to $0.010 [\mu g \cdot L^{-1}]$ for benzotriazoles, and the highest from 0.016 to $0.936 [\mu g \cdot L^{-1}]$ for glycols respectively.

Figure 3 provides an example of the chromatograms obtained from the analysis of runoff water samples collected from local and international airport in Poland and from international airport in the United Kingdom, using extraction techniques A1-A5.

- 197
- 198 **3. Results**

In order to illustrate the possibilities of using five different extraction techniques (A1-A5), which are the main stage of the developed analytical procedures for the determination of PAHs, BTs and glycols, the results of the research on airport runoff water samples collected in the areas of three airports (international airport, local airport in Poland and international airport in United Kingdom) were presented. During the period from autumn 2011 to spring 2012, 207 runoff water samples were collected (125 samples -national airport PL, 55 samples-local airport PL, 27 samples-international airport UK).

In the samples collected in the areas of three airports, which are the subject of this research, compounds, which represent a group of pollutants originated from fuel combustion (PAHs), were detected and analyzed. Figure 4 presents the results of the concentration levels of individual PAH analytes determined in runoff water samples collected in the areas of airports (international airport PL, local airport PL). The highest concentrations of PAH analytes were noted in the samples collected from the international airport PL. Compounds whose concentrations were the highest, regardless of the airport, were chrysene and phenanthrene.

213 Histograms presented in the Figure 5 are the source of information on the concentrations 214 of individual analytes from the group of benzotriazoles determined in runoff water samples 215 collected in the area of the airports (Polish international, local airport and British international 216 airport). The quantitative analysis showed that the concentration levels of BTs in runoff water 217 samples collected from international and national Polish airports were significantly greater 218 compared with other locations. The highest concentrations of 5-Me-1H-1H-BT and BT were 219 determined in the sample taken from Polish international airport from the measuring point, where 220 de-icing operations are performed (89.3 µg·L-1 and 29.1 µg·L-1 respectively). In general, the 221 highest concentration levels of compounds that represent a group of pollutants originated from 222 anti-corrosive substances (e.g. BTs) were determined in the samples taken from aircraft de-icing 223 places, machinery parks and technical roads in the airports. Relatively low concentration levels of 224 BTs were determined in the samples taken from the area of British international airport.

225 At all airports under investigation, compounds, which represent a group of pollutants 226 originated from anti-icing substances (e.g. glycols) were detected and analyzed. This is illustrated 227 by the data shown in Figure 6 concerning the concentrations of glycols in runoff water samples 228 collected from the area of airports (international and local polish airport and also British 229 international airport). Generally, the concentration levels of glycols in the samples collected from 230 international British airport were much higher (254 mgL-1) compared with other locations. 231 Among the compounds from the group of glycols, the highest concentrations were reported for 232 ethylene glycol in all cases. High concentration levels of the compounds from the group of 233 glycols were determined mainly in the samples collected from the areas, where aircraft de-icing 234 processes are performed, as well as within the parking places (where de-icing operations are 235 carried out in the smaller airports) and airport technical roads (transmission and scattering of de-236 icing fluids by maintenance vehicles).

237 **4. Discussion**

238 Intensification of air transport, i.e. the increase in air traffic at the airports and the 239 development of airport network, causes that anthropogenic impact on the environment is more 240 and more significant. There is no doubt that it is necessary to conduct a comprehensive control of 241 the composition of the airport runoff water samples. The study of pollutants present in specific 242 environmental samples, such as runoff water samples from the area of the airports, allows a better 243 understanding and characterization of the occurrence of xenobiotics in the environment. Only 244 widely spread monitoring of such samples can allow to obtain data, which will be the basis for 245 evaluating the intensity of the airport activity on the inanimate nature and living organisms and 246 subsequently to prepare strong foundations for the management of waste produced by the 247 airports.

248 In order to estimate the amount of individual xenobiotics released to runoff waters and to 249 investigate their environmental fate appropriate analytical procedures are required. The most 250 difficult step in the analytical procedure (during which significant errors can be made) used for 251 the determination of the described in this work groups of compounds that represent the pollution 252 from fuel combustion products, anti-corrosive compounds, and de-icing compounds in airport 253 runoff waters is the sample preparation step. It is due to the diversity of the compounds present in 254 the samples collected from the airport platform, a diverse matrix composition, the possibility of 255 interference associated with the presence of components with similar physico-chemical 256 properties, the lack of the reference materials necessary to ensure an adequate level of assurance 257 and quality control of the analytical measurements).

In this study, five different versions of sample preparation using extraction techniques, such as: LLE and SPE, were tested. This work also presents complete analytical procedures suitable for the determination of the analytes from the group of PAHs, BTs and glycols present in

261 the airport runoff water samples. The development of the procedure for determining some of the 262 most toxic xenobiotics in runoff waters was used to analyze real samples collected from three 263 airports located on the Polish territory and in the United Kingdom. In all examined runoff water 264 samples collected from the airports, the presence of compounds from the group of PAHs and 265 glycols was confirmed. Moreover, the presence of BTs was observed in the majority of the 266 samples. Runoff water samples collected from the areas of Polish and British international 267 airports as well as local airports had similar qualitative composition, and quantitative 268 composition of the analytes was very diverse, which depends on the emission of pollutants 269 generated by the airport, but also on the meteorological conditions in a given area.

This type of research is conducted in only a few reputable scientific institutions. However, it can be confidently stated that runoff waters from the area of the airports are attracting greater interest as a source of information about the potentially negative impact of the rapidly increasing airport activity on the state of the environment.

274

275 Acknowledgements

The authors would like to thank the Ministry of Science and Higher Education for the grant awarded within the framework of the Project No. 0528/IP3/2011/71 for financing this investigation. The authors express their gratitude to Eng. Krzysztof Suchomski for his help in preparing the graphic side of this publication.

280 **4. References**

[1] M. Pitt, A. Smith, Journal of Air Transport Management, 9 (2003) 103-111.

[2] M. Switzenbaum, S. Veltman, M.D. Wagoner, T. Schoenberg, Chemosphere, 43 (2001) 10511062.

[3] L. Luther, in: S. Environmental Policy Analyst Resources, and Industry Division (Ed.), 2007.
[4] S.R. Corsi, G.R. Harwell, S.W. Geis, D. Bergman, Environmental Toxicology and Chemistry,
25 (2006) 2890-2900.

- [5] S. Barash, J. Covington, C. Tamulonis, in: U.S.E.P. Agency (Ed.), Washington, 2000.
- 288 [6] A. Kiss, E. Fries, Environ Sci Pollut Res, 16 (2009) 702-710.
- 289 [7] S. Weiss, J. Jacobs, T. Reemtsma, Environ Sci Technol, 40 (2006) 7193-7199.
- [8] M.S. Schwitzenbaum, S. Vetman, T. Schoenberg, C.M. Durand, D. Mericas, B.W. No.173, in,1999.
- 292 [9] T. Reemtsma, M. Ulf, U. Duennbier, M. Jekel, Water Res., 44 (2010) 596-604.
- [10] D. Pillard, D. DuFrense, Arch Environ Contam Toxicol, 37 (1999) 29-35.
- 294 [11] D. Pillard, J. Cornell, D. Dufresne, M. Hernandez, Water Res., 35 (2001) 557-560.
- [12] E. Hoffman, G.L. Mills, J.S. LAmiter, J.G. Quinn, Environ Sci Technol, 18 (1984) 580-587.
- [13] C. Sun, J. Fitzpatrick, S. Goebel, J. Loos, in, University of Missouri—Columbia, 2011.
- [14] G.A. Burton, R.E. Pitt, Stormwater Effects Handbook: A CRC/Lewis, London, 2002.
- 298 [15] D.M. Revitt, R.B.E. Shutes, N.R. Llewellyn, P. Worrall, Wat. Sci. Tech, 36 (1997) 385-390.
- [16] G.D. Breedveld, R. Roseth, M. Sparrevik, T. Hartnik, L.J. Hem, Water, Air, Soil Pollut, 3(2003) 91-101.
- 301 [17] V. Barcan, E. Kovnatsky, A. Shylina, Atmospheric Environment 34 (2000) 1225-1231.
- 302 [18] D.M. Ramakrishna, T. Viraraghavan, Water, Air and Soil Pollution, 166 (2005) 49-63.
- 303 [19] S. Grant, N. Rekhi, N. Prise, R. Reeves, in: C.D.o. Transportation (Ed.), Caltrans,
 304 Sacramento, 2003.
- 305 [20] A. Sulej, Ż. Połkowska, J. Namieśnik, Critical Reviews in Science and Technology, 42
 306 (2012) 1691-1734.
- 307 [21] A.M. Sulej, Ż. Połkowska, J. Namieśnik, Critical Reviews in Analytical Chemistry, 41
 308 (2011) 190–213.
- [22] D. Cancilla, J.C. Baird, S.W. Geist, S. Corsi, Environmental Toxicology and chemistry, 22
 (2003) 134-140.
- 311 [23] D.A. Cancilla, J. Martinez, G.C.v. Aggelen, Environ. Sci. Technol., 32 (1998) 3834–3835.
- 312 [24] H.K. French, S.E.A.V.d. Zee, A. Leijnse, J. Contam. Hydrol, 49 (2001) 23–48.
- 313 [25] W. Giger, in: E.a. research (Ed.), Swiss Federal Institute of Aquatic Science and 314 Technology, Stresa, 2006.
- 315 [26] J.M. Strong-Gunderson, S. Wheelis, S.L. Carroll, M.D. Waltz, A.V. Palumbo, in, 1995.
- [27] S.R. Corsi, S.W. Geis, J.L.-. Rosales, C.P. Rice, R.J. Sheesley, G.G. Failey, D. Cancilla,
 Environ. Sci. Technol., 40 (2006) 3195-3202.
- [28] S.R. Corsi, Si Wi Geis, G. Bowman, G.G. Failey, T.D. Rutter, Environ. Sci. Technol., 43
 (2009) 40-46.
- [29] S.R. Corsi, D.J. Zitomer, J. Field, D.A. Cancilla, Environ. Sci. Technol, 37 (2003) 4031 4037.
- [30] J.W. Readman, G. Fillmann, I. Tolosa, Ja Bartocci, J.P. Villeneuve, C. Catinni, L.D. Mee,
 Marine Pollution Bulletin 44 (2002) 48–62.
- 324 [31] P. Baumard, H. Budziński, P. Garrigues, Marine Pollution Bulletin, 36 (1998) 577-586.
- [32] J.A. Leerdam, A.C. Hogenboom, M.M.v.d. Kooi, P.d. Voogt, International Journal of Mass
 Scectrometry, 282 (2009) 99-107.
- 327 [33] Y.-S. Liu, G.-G. Ying, A. Shareef, R.S. Kookana, Journal of Chromatography A, 1218
 328 (2011) 5328-5335.
- 329 [34] D.S. Hart, L.C. Davis, L.E. Erickson, T.M. Callender, Microchem. J., 77 (2004).
- 330 [35] D.H. Zitomer, in, Marquette University, Milwaukee, 2001, pp. 38.
- [36] U.S.D. Transportation, in, Federal Aviation Administration 2010.
- 332 [37] P. Popp, C. Baurer, B. Hauser, P. Keil, L. Wennrich, J. Sep. Sci., 26 (2003) 961-967.
- [38] M. O'Donnell, in: U.S.D.o. Transportation (Ed.), Federal Aviation Administration, 2008.

- 334 [39] J. Pawliszyn, Handbook of sample preparation, A John Wiley & Sons, Inc., Publication,
- 335 New Yearsy, 2010.
- [40] W. Zhang, S. Zhang, D. Yue, C. Wan, Y. Ye, X. Wang, Environ Toxicol Chem., 27 (2008)
 31-37.
- 338 [41] K. Lamprea, V. Ruban, Environ Technol., 32 (2011) 1141-1149.
- 339 [42] C. Harris, E. Routledge, C. Schaffner, J. Brian, W. Giger, J. Sumpter, Environ Toxicol
- 340 Chem., 26 (2007) 2367-2372.
- 341 [43] Ż. Polkowska, K. Skarżyńska, T. Górecki, J.Namieśnik, Atmospheric Environment, 36
- 342 (2002) 361-369.
- [44] M. Korhonen, A. Kiviranta, R. Ketola, Toxicological and Environmental Chemistry, 66
 (1998) 37-45.
- 345 [45] K. Levsen, S. Behnert, B. Prieb, M. Svoboda, H. D. Winkeler, J. Zietlow, Chemosphere, 21
- 346 (1990) 1037–1061.
- 347 [46] W.F. Rogge, L. Hildemann, M.A. Mazurek, G.R. Cass, B. R.Simoneit, Environmental
- 348 Science and Technology, 27 (1993) 1892-1904.
- 349 [47] E. Manoli, C. Samara., Trends in Analytical Chemistry, 6 (1999) 417–428.
- 350 [48] D. Massart, B. Vandeginste, S. Deming, Y. Michotte, L. Kaufman, Chemometrics: a
- 351 textbook, Elsevier, Amsterdam, 1988.
- 352
- 353

Type of pollutant	Origin of pollution	Major toxicants
Fuel combustion	• combustion of fuels	• PAHs
products	• aircraft, vehicle maintenance	• PCB
	operations	• phenols, formaldehyde
	• fuelling operations	• benzene
	• engine testing operations	
Anti-corrosion	• aircraft de-icing anti-icing	Benzotriazoles
pollutants	fluid additives (ADAFs)	• 4-methyl-1-H-benzotriazole
	• engine coolants and oils in	(4-MeBT)
	automobiles	• 5-methyl-1-H-benzotriazole
	• industrial cooling systems	(5-MeBT)
	• plastic stabilizers	• 1H-benzotriazole (1-MeBT)
De/anti-icing	de/anti-icing operations	Glycols:
chemical wastes		• propylene glycol (PG)
		• ethylene glycol (EG)
		• diethylene glycol (DEG)

1 Table 1 Major sources of emission and xenobiotics generated during airport operations

Table 2 Reagents and apparatus used in this research study

	Apparatus and reagents						
Apparatus	Sample preparation	Centrifuge shaker (Conbest, ELMI, Poland), Vacuum set of SPE-12G [™] (J.T. Baker, Poland),					
	Finally determination	Agilent 7890A gas chromatograph coupled with mass spectrometer Agilent 5975C,					
Solvents	Methanol, <i>n</i> -heksane, Dic	chloromethane (Lichrosolv, Merck, Germany);					
SPE columns	Strata C-18E (Phenomene	ex, USA),					
	ENVI-Carb Plus (Sigma-	Aldrich, USA),					
Standard	Mix of 16 PAHs (2000 μg·mL ⁻¹ in dichloromethane, Supelco, USA),						
solutions	Mix of benzotriazoles (1000 μ g·mL ⁻¹ in methanol, Sigma-Aldrich, Germany),						
	Mix of glycols (1000 µg·mL ⁻¹ in methanol, Sigma-Aldrich, Germany)						
Other	Deionized water Milli-Q (Millipore Corporation, USA), Nitrogen (purity 99,99%, Poland)						

- 21 Table 3 The characteristics of the places of sample collection of airport runoff waters

Sample number	Locations of sample collection					
number						
Airport	International PL	Local PL	International UK			
1	influent of a river	vicinity of an airport terminal	de-icing area (1)			
2	effluent of a river	de-icing area	a river in the vicinity of the airport			
3	municipal water catchment area	machinery stock, parking places	de-icing area (2)			
4	CARGO water catchment area	runway	de-icing area (3)			
5	airport ramp	parking places	de-icing area (4)			
6	car park	the periphery of an airport	a road near the airport			
7	de-icing area	car park	-			
8	airport ramp	-	-			

26 Table 4 Comparison of analytical procedures used for the determination of PAHs, BTs and glycols.

Analytical procedures/Analytes	Pre-treatment method	Analytical method	Recovery (%)	RSD (%)	Advantages	Disadvantages		
A1-A3 PAHs (16analytes from 1	A1-A3 PAHs (16analytes from PAHs group)							
A1	LLE	GC-MS	60 - 80	6 – 12	Simplicity of the used	Large amounts of solvent needed		
					apparatus	Large amounts of a sample needed		
					Suitability for highly	Time-consuming and labor-intensive		
Tom most A2 A3					contaminated samples			
	C18-SPE				Smaller amount of a sample			
			72 – 101	1.8 - 5.2	needed	Time-consuming		
→ A 3 00	C18-SPE		78 – 102	1.5 – 5.5	High LODs	Time-consuming		
A4 BENZOTRIAZOLES	<u> </u>	<u> </u>	<u> </u>	<u> </u>	1			
₫-H-benzotriazole	C18-SPE	GC-MS	68	12	High recovery	High RSD		
ethyl-1H-benzotraizole			102	8.1		Relatively high consumption of		
ethyl-1H-benzotraizole	_		88	7.2	-	solvent		
SLYCOLS	I	I	I	I				
pylene glycol	Envi-CarbPlus-	GC-MS	86 - 94	2.5 - 3.1	Small amount of a sample	Mechanical clogging of the pores by		

Ethylene glycol	SPE	70 -78	6.7 – 7.7	High recovery	runoff water samples
Diethylene glycol		89 - 96	8.6 - 8.9		

Table 5 Factor loadings and explained variance of varimax rotated factors

	Factor 1	Factor 2	Factor 3	Factor 4
Naphthalene	-0.12	0.75	0.25	0.20
Acenaphthylene	0.39	0.45	-0.10	0.53
Acenaphthene	0.03	0.90	0.08	0.09
Fluorene	0.91	0.06	0.06	-0.11
Phenanthrene	0.28	0.43	0.45	0.36
Anthracene	0.11	0.63	0.11	-0.14
Fluoranthene	0.52	-0.09	0.69	0.37
Pyrene	0.90	-0.04	0.35	0.18
Benz[a]anthracene	0.08	0.03	0.85	0.01
Chrysene	0.10	0.18	0.92	0.21
Benzo[b]fluoranthene	0.86	0.02	0.03	0.31
Benzo[k]fluoranthene	0.59	-0.00	0.03	0.72
Benzo[a]pyrene	0.21	0.06	0.47	0.75
Indeno[1,2,3-c,d]pyrene	0.06	0.16	0.29	0.86
Dibenz[a,h]anthracene	-0.12	0.62	-0.13	0.56
Benzo[g,h,i]perylene	0.13	0.42	0.51	0.61
Eigenvalue	3.34	2.79	3.06	3.30
Variance explained [%]	21	17	19	21

Table 6 Basic parameters of the analytical procedure for PAH determination in airport runoff water samples based on the application of LLE, SPE and GC-MS techniques (obtained from the analysis of the samples of standard solutions).

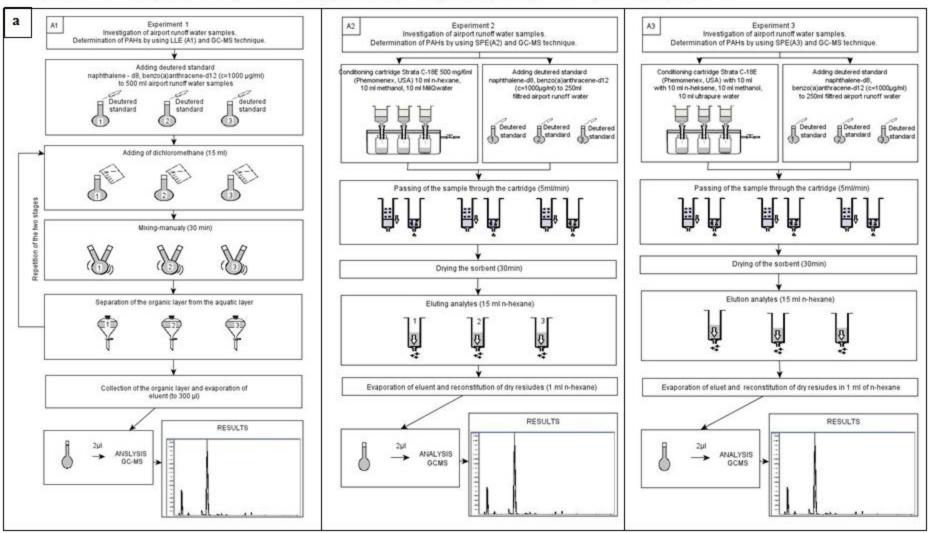
Conditions of the analysis							
Analyte	PAHs	Benzotriazoles	Glycols				
Sample preparation technique	LLE and SPE	SPE	SPE				
Element of the measurement	GC-EI-MS	GC-EI-MS	GC-EI-MS				
system							
Gas chromatograph	Agilent 7980A	Agilent 7980A	Agilent 7980A				
Detector	Agilent 5975C	Agilent 5975C	Agilent 5975C				
Detector working mode	monitoring of	monitoring of selected	monitoring of selected				
	selected ions	ions	ions				
Temperature of the source of		230 °C					
ionization							
Quadrupole temperature		150 °C					
Energy of the electron stream		70eV					
Chromatographic column	ZB-5MS;		SPB-1000				
	30m x 0,25mm; 0,25µ	m stationary phase	30m x 0,25mm; 0,25µm				
Stationary phase	5/95 phenyl/polydimet	thylsiloxane	modified polyethylene				
			glycol				
Pressure of the carrier gas (He)	7,07 psi	8,80 psi	7,65 psi				
Flow rate of the carrier gas		$1 \text{ ml} \cdot \text{min}^{-1}$					
Temperature of injection port	295°C	260 °C	220 °C				
Connection temperature	295°C 260 °C 220 °C						
Working mode of injection port	splitless	1					
Temperature programme	40-120 °C (40 °C/min)	70°C (2 min)	50-200 °C (8 °C/min)				
	120-280 °C (5°C/min)	70-275°C (14 °C/min)	200 °C (1min)				

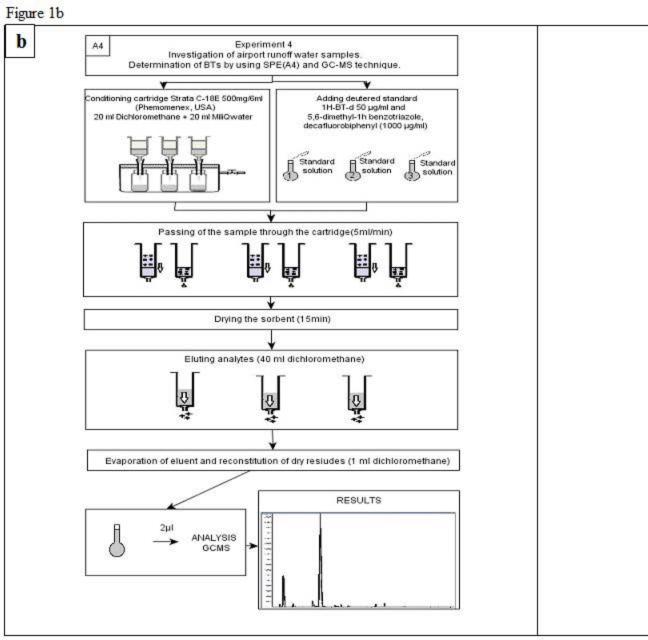
		275°C (2	2 min)			
Volume of injection	2 μl			1 µl		
Time of the analysis	46 minutes	19 minutes20 minutes				
Number of the analytes	16		3			
	Metrological	characte	eristics			
	Polycyclic arom	atic hydr	ocarbons			
Analyte	Calibration curve eq	uation	Regressio	n LOD	LOQ	
			coefficient,	$\mathbf{R}^2 \qquad [\mu \mathbf{g} \cdot \mathbf{L}^{-1}]$	[µg·L ⁻¹]	
Naphthalene	$y=5.00\cdot10^{6}x-6.44\cdot10^{4}$		0.982	0.013	0.040	
Acenaphthylene	$y=2.00\cdot10^{6}x+9.96\cdot10^{3}$		0.997	0.0003	0.001	
Acenaphthene	$y=4.00\cdot10^{6}x+2.98\cdot10^{3}$		0.998	0.001	0.004	
Fluorene	$y=3.00\cdot10^{6}x-3.26\cdot10^{4}$		0.995	0.013	0.030	
Phenanthrene	$y=6.42\cdot10^{5}x-8.31\cdot10^{3}$		0.993	0.043	0.130	
Anthracene	$y=4.22\cdot10^{5}x+3.83\cdot10^{4}$		0.994	0.017	0.051	
Fluoranthene	$y=8.81\cdot10^{5}x+6.89\cdot10^{3}$		0.996	0.001	0.004	
Piren	y=8.66·10 ⁴ x+7.57·10 ⁴		0.985	0.0005	0.002	
Benz[a]anthracen	$y=8.84\cdot10^{4}x+2.86\cdot10^{3}$		0.997	0.006	0.019	
Chrysene	$y=8.90\cdot10^4x+2.01\cdot10^3$		0.998	0.004	0.011	
Benzo[b]fluoranthene +	$y=2.29\cdot10^{5}x+1.11\cdot10^{4}$		0.997	0.057	0.170	
benzo[k]fluoranthene						
Benzo[a]pyrene	$y=1.49\cdot10^{5}x+2.10\cdot10^{3}$		0.994	0.053	0.160	
Indeno[1.2.3-cd]pyrene	$y=1.53\cdot10^{5}x+1.64\cdot10^{2}$		0.998	0.057	0.170	
	Benzotria	zoles				
1H-benzotriazole	$y=2.92\cdot10^{5}x+2.84\cdot10^{4}$		0.983	0.010	0.030	
4-Me-1H-benzotriazole	$y=5.98\cdot10^{5}x-3.00\cdot10^{6}$		0.962	0.010	0.030	
5-Me-1H-benzotriazole	$y=4.44\cdot10^{5}x+1.44\cdot10^{5}$		0.942	0.0003	0.001	
	Glyco	ls				

Ethylene glycol	$y=4.88\cdot10^{5}x-4.00\cdot10^{6}$	0.970	0.189	0.567
Diethylene glycol	$y=1.00\cdot10^{6}x+2.41\cdot10^{5}$	0.986	0.016	0.047
Propylene glycol	$y=8.62\cdot10^{5}x-5.00\cdot10^{6}$	0.977	0.936	2.81

Figure 1a

The schematic presentation of the analytical procedures for the determination of PAHs in the airport runoff water samples by means of : LLE (A1), SPE (A2), and SPE (A3) (Figure 1 a) and SPE (A4), SPE (A5) techniques (Figure 1b) at the step of sample preparation.





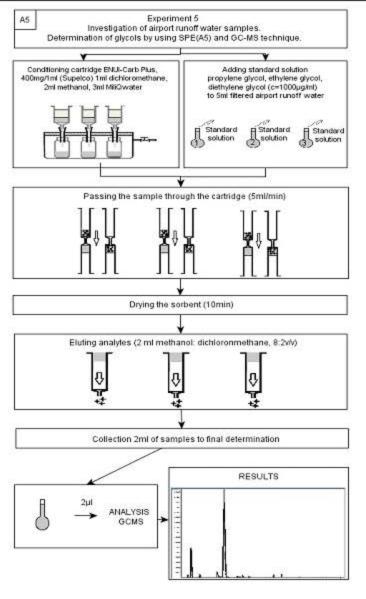


Figure 2

Plot of sample scores of the first and second factor identified (Figure 2a), the first and third factor identified (Figure 2b), and the first and fourth factor identified (Figure 2c) by principal component analysis with varimax rotation

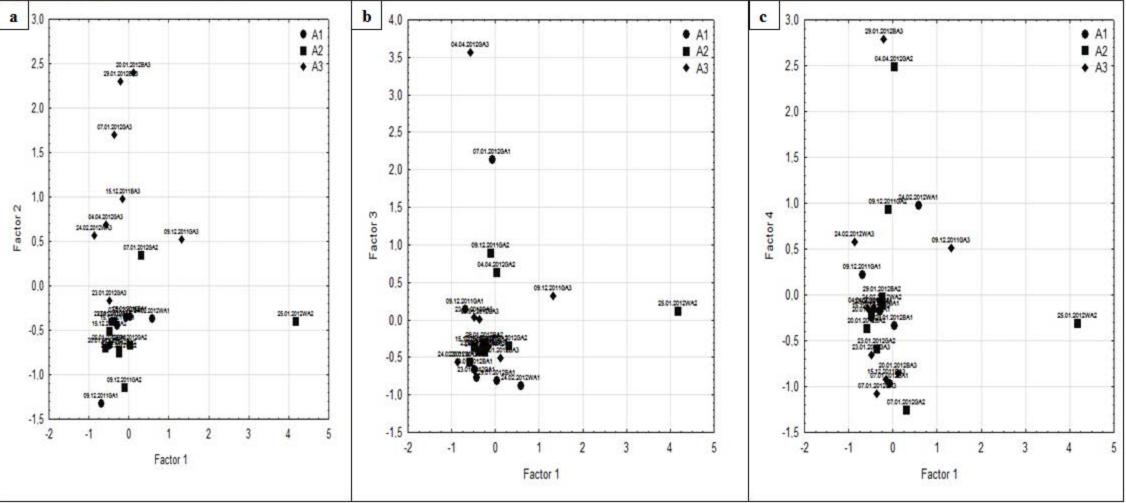
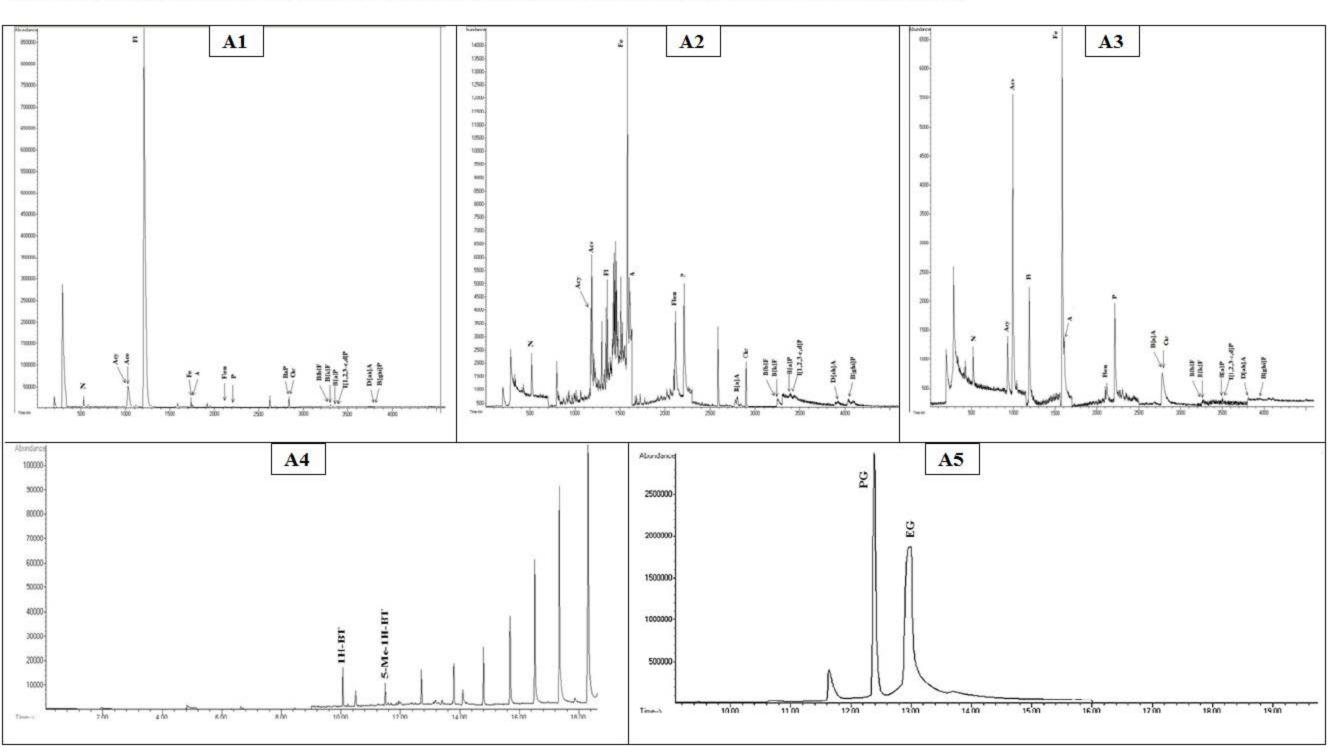


Figure 3 Chromatograms obtained as a result of the analysis of the airport runoff water samples (samples were prepared by means of A1-A5 (LLE and SPE) techniques).



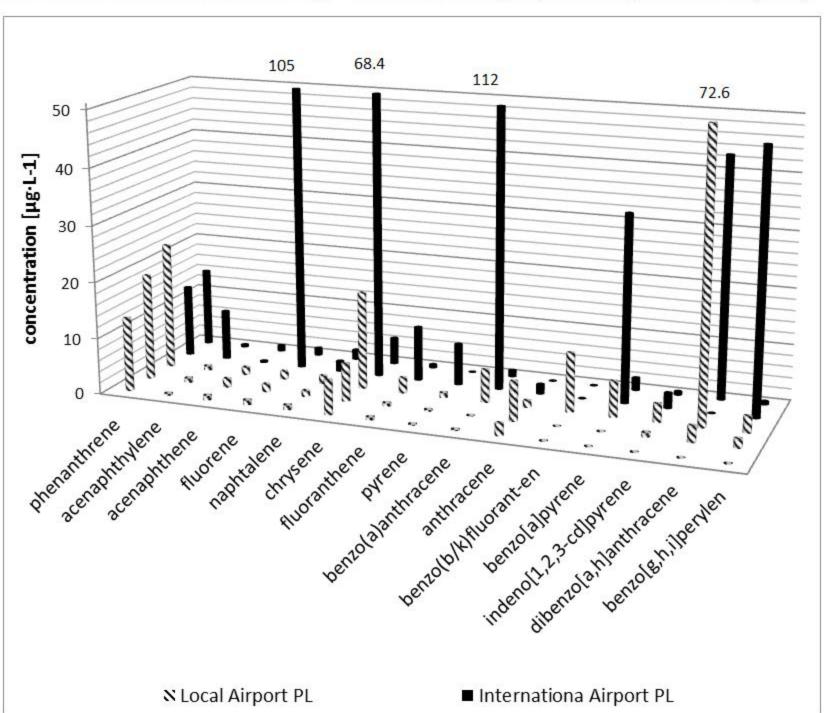


Figure 4 PAH concentrations determined in runoff water samples collected in the area of airports (international airport PL and local airport PL).

Figure 5

Concentration levels of benzotriazoles determined in airport runoff water samples collected from international airport PL (Figure 5a), local airport PL (Figure 5b), and international airport UK (Figure 5c).

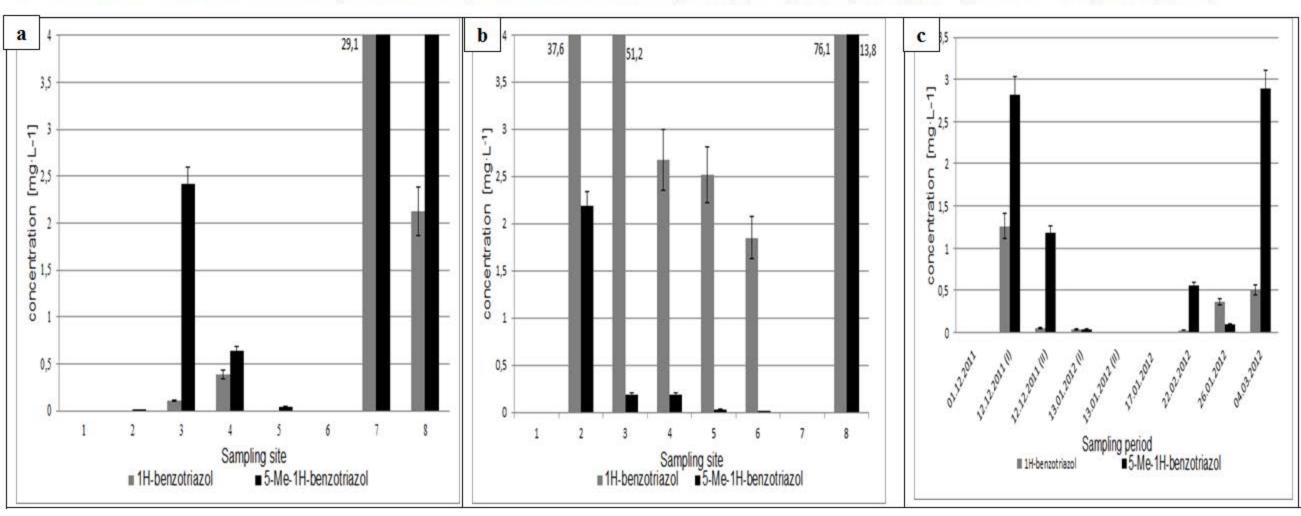


Figure 6

Concentration levels of individual glycols determined in the airport runoff water samples collected from international airport PL (Figure 6a), local airport PL (Figure 6b), and international airport UK (Figure 6c).

