Superconductivity at 2.2 K in the layered oxypnictide La₃Ni₄P₄O₂

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We report the observation of superconductivity in La₃Ni₄P₄O₂ at 2.2 K. The layer stacking in this compound results in an asymmetric distribution of charge reservoir layers around the Ni₂P₂ planes. The estimated Wilson ratio, $R_W \approx 5$, indicates the presence of strongly enhanced normal-state susceptibility, but many of the basic superconducting characteristics are conventional. The estimated electronic contribution to the specific heat, $\gamma \approx 6.2$ mJ mol-Ni⁻¹ K⁻², is about 2/3 of that found in layered nickel borocarbide superconductors.

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Like the high- T_c copper oxides, the oxypnictide superconductors²⁻⁴ are based on square metal layers. Several structural classes of these superconductors (with $X-M_2-X$ layers, where M=Fe or Ni, and X=P, As, or Se), are currently known. The simplest is based on β -FeSe,⁵ which consists of Se-Fe₂-Se layers only, and can be considered as the parent compound of the family. A related compound, Li_rFeAs, also superconducts.^{6,7} Others, typified by BaNi₂P₂ (Ref. 8) and K-doped BaFe₂As₂, are derived from the β -FeSe-like parent by inserting layers of alkali earth ions between $X-M_2-X$ layers. Superconducting LaFeAsO and La-NiPO (Refs. 2 and 3) are derived instead by inserting whole La-O₂-La layers between the $X-M_2-X$ layers. Here we show that La₃Ni₄P₄O₂, a compound that has a mixed-layer structure between LaNi₂P₂ and LaNiPO, is superconducting with a T_c of 2.2 K. The stacking of inequivalent charge reservoir layers in La₃Ni₄P₄O₂ leads to an asymmetric distribution of ionic layers around the Ni₂P₂ planes, a structure-derived electronic feature that is not observed in any other known member of the superconducting oxypnictides.

Polycrystalline La₃Ni₄P₄O₂ samples were prepared from Ni₅P₄, elemental P, and dry La₂O₃. ¹⁰ Physical property characterization was performed on a Quantum Design physical property measurement system (PPMS). Neutron-diffraction data were collected at the NIST Center for Neutron Research on the BT-1 powder neutron diffractometer. 11 The crystal structure was determined by a Rietveld fit^{12,13} to powder neutron-diffraction data. The diffracted peak positions and intensities were well fit by a tetragonal Pr₃Cu₄P₄O₂-type crystal structure. 14 The fractional occupancies of each crystallographic site were independently refined and, within two standard deviations, were all found to be equal to one. Therefore the structure can be considered stoichiometric, and for the final refinement, all site occupancies were fixed at unity. The refined structural parameters are summarized in Table I. Small quantities of two impurity phases are present in the best preparations; 10.0(2)% of LaNiPO and 6.1(2)% LaNi₅P₃ were included in the final fit for the sample characterized in this study.

The crystal structure of La₃Ni₄P₄O₂ is made from covalently bonded P-Ni₂-P layers, the common structural motif seen throughout the superconducting oxypnictide family. The M-X bond lengths and M-X-M bond angles within these covalent layers are under discussion as important chemical influences on the electronic properties in this family of compounds (see, e.g., Ref. 15), as is X-X bonding. 16 These layers are separated in the new compound by an alternating pattern of layers of La and La-O₂-La. Figure 1 shows the crystal structure, which is an ordered structural mixture of LaNiPO (Ref. 3) and LaNi₂P₂, ¹⁷ and its relationship to those phases and the parent β -FeSe structure type. These compounds form a homologous series with the formula $(La_2O_2)_m[La/Ba]_n(Ni_2P_2)_{n+m}$, and, including $La_3Ni_4P_4O_2$, the m,n pairs (1,0), (0,1), and (1,1) are known to support superconductivity. In Fig. 1, the La³⁺ and O⁻² ions are drawn using ionic radii and the M and X atoms are drawn using covalent radii. This rendering reflects a chemically interesting aspect of the oxypnictide superconductors—the intimate intergrowth of highly ionic layers such as La3+ and $(La_2O_2)^{2+}$ with highly covalent $X-M_2-X$ layers such as Ni_2P_2 or Fe₂As₂. These ionic layers are the "charge reservoirs;" completely analogous to those in the layered high- T_c copper

TABLE I. Refined structural parameters for La₃Ni₄P₄O₂ at 298 K. Space group *I*4/*mmm* (139), a=4.0107(1) Å, c=26.1811(9) Å. Calculated density 6.36 gm cm⁻³. $U_{\rm iso}$ =thermal vibration parameters in 10⁻² Å². Figures of merit: goodness of fit (χ^2)=1.073, weighted profile residual ($R_{\rm wp}$)=4.87%, profile residual (R_p)=3.98%, and residual on structure factors [$R(F^2)$]=5.45%.

$La_3Ni_4P_4O_2$ $T=298$ K			
Atom	$U_{ m iso}$	Position	Z
La1	0.88(5)	2a (0,0,0)	
La2	0.88(5)	4e (0,0,z)	0.7013(1)
Ni	1.15(4)	8g(0,1/2,z)	0.09278(7)
P1	0.84(7)	4e (0,0,z)	0.1335(2)
P2	0.84(7)	4e (0,0,z)	0.5493(2)
0	1.2(1)	4d (0,1/2,1/4)	

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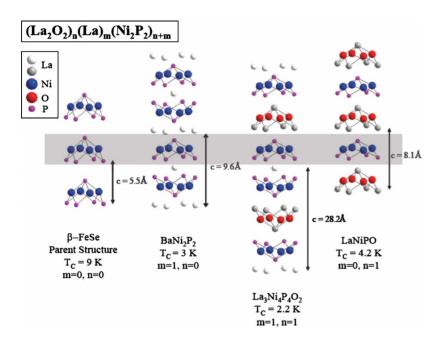


FIG. 1. (Color online) La₃Ni₄P₄O₂ is a member of a series of compounds built by inserting charged ions or layers between M_2X_2 sheets of the parent β -FeSe structure. All the oxypnictide superconductors are members of this family. The La³⁺ and O⁻² ions are drawn with ionic radii and the M and X atoms are drawn with covalent radii.

oxides. What is unique about La $_3$ Ni $_4$ P $_4$ O $_2$ among the oxypnictide superconductors is that the distribution of charge reservoir layers around the Ni $_2$ P $_2$ plane is highly asymmetric; i.e., the sequence of layers is [La 3 +]-[P-Ni $_2$ -P]-[La-O $_2$ -La] 2 +. Thus, the Ni $_2$ P $_2$ layer is sandwiched between two different ionic layers—an arrangement that should lead to the presence of a local electric-dipole field across the superconducting plane.

Figure 2 shows the superconducting transition for La₃Ni₄P₄O₂ characterized by magnetic susceptibility (H_{DC} = 10 Oe). The critical temperature (T_c) is about 2.2 K. The minor diamagnetism visible in the temperature range 2.5 K to 4 K is associated with the presence of the 10% LaNiPO impurity phase. As is the usual case, pinning of vortices produces a smaller field-cooled (FC) signal compared to zerofield cooling (ZFC). The diamagnetic response in ZFC measurements is substantially larger than $1/4\pi$, which we attribute to demagnetization effects of the rectangular-shaped sample. To estimate the demagnetization factor (d), measurements of the low-field magnetization were made at 1.9 K, shown in Fig. 3. Assuming that the initial linear response to fields less than 50 Oe is perfectly diamagnetic, that is, that the slope dM/dH is $-1/4\pi$, we obtain a demagnetization factor that is consistent with the sample's shape. Applying this demagnetization factor gives the maximum ZFC susceptibility in Fig. 2 of approximately $-1.1(1/4\pi)$. As shown in the inset of Fig. 3, M(1.9 K) begins to deviate from a fitted linear dependence on H at a field $H^* \approx 60$ Oe, giving a lower critical field, 18 taking into account the demagnetization factor, of $H_{C1}(1.9 \text{ K}) = H^*/(1-d) \approx 90 \pm 10 \text{ Oe.}$ The estimated uncertainty reflects criteria that H^* is either the field above which M first deviates from the fitted line or the field where M deviates by 2.5% above the fitted curve. On the basis of this H_{C1} at 1.9 K and T_c in zero field, a crude estimate of the zero-temperature lower critical field is about 520 Oe, which would imply a Ginzburg-Landau superconducting penetration depth of approximately λ_{GL} of $550 \pm 50\,$ Å. ¹⁸

The inset of Fig. 2 shows the temperature dependence of

the estimated normal-state magnetic susceptibility. Low-field (H=1 kOe) measurements indicated the presence of small amounts of ferromagnetic impurity with a low coercive field, likely elemental nickel, in the best samples. This makes the usual low-field measurement of susceptibility invalid due to the relatively high contribution of the ferromagnetic impurity to the measured magnetization at low fields. Therefore, to approximate the intrinsic susceptibility of La₃Ni₄P₄O₂, the difference in magnetization ΔM between applied fields of 50 and 40 kOe, well above the coercive field of the ferromagnetic impurity, was employed to estimate the susceptibility $\chi = \Delta M / \Delta H$. Above 250 K La₃Ni₄P₄O₂ exhibits temperatureindependent Pauli paramagnetism with $\times 10^{-4}$ emu mol⁻¹. Below 250 K, $\chi(T)$ increases with decreasing temperature and reaches maximum at about 14 K. This may reflect the existence of a magnetic transition in this phase, but the presence of LaNiPO and LaNi₅P₃ impurities

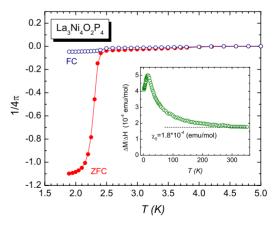


FIG. 2. (Color online) Superconducting transition measured in zero-field-cooled (closed circles) and field-cooled mode (open circles) under applied magnetic field 10 Oe. Inset of the figure represents magnetic susceptibility $(\Delta M/\Delta H)$ in the wide temperature range. $\chi = \Delta M/\Delta H$ was calculated as the difference in magnetization ΔM between applied fields of 50 and 40 kOe (see text).



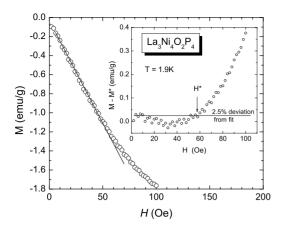


FIG. 3. Magnetization (M) versus applied magnetic field (H) at constant temperature 1.9 K. The solid line corresponds to linear relation $(\sim H)$ for H < 50 Oe. The inset shows deviation from a fitted linear dependence on H at a field H^* (obtained from the fit between 5 and 50 Oe),

precludes more detailed analysis at this time.

Specific-heat (C_p) measurements performed at 0 and 90 kOe magnetic fields are shown in Fig. 4. The data in zero field (closed circles) show an anomaly at T_c =2.3 K, close to the T_c determined by susceptibility measurement. (There is also a small anomaly near 4 K due to the LaNiPO impurity.) The normal-state specific heat measured at $\mu_0 H=9$ T, shown by open circles, was fitted to $C_p = \gamma T + \beta T^3 + A/T^2$, where the first, second, and third terms represent electronic, lattice, and nuclear Schottky contributions, respectively. The fit to these data allows us to estimate γ =6.2 mJ mol-Ni⁻¹ K⁻², β =0.14 mJ mol-Ni⁻¹ K⁻⁴, and \hat{A} =0.49 mJ mol-Ni⁻¹ K. The value of γ is smaller than those found in the layered nickel borocarbide superconductors, e.g., LuNi₂B₂C (8.9 mJ mol-Ni⁻¹ K⁻²), ¹⁹ which also have significantly higher T_cs (16.5 K for LuNi₂B₂C). Given both γ and χ_0 , the Wilson ratio can be estimated for La₃Ni₄P₄O₂, yielding $R_W = \frac{(\pi k_B)^2 \chi_0}{3 \mu_{eff}^2 + 2 \lambda_0} \approx 5 \left[\mu_{eff} = 2.76 \times 10^{-21} \right]$ erg/Oe was cal-

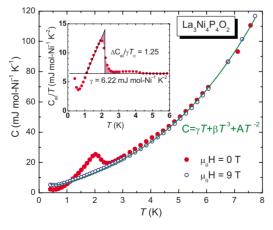


FIG. 4. (Color online) Heat capacity data under 0 T and 9 T applied magnetic field. The green line represents a fit to the data (see text). The inset shows electronic specific heat $(C_{\rm el.}/T)$ in the low-temperature range. The entropy conservation construction gives $\Delta C_{\rm el}/\gamma T_c$ =1.25, underestimated due to the presence of the impurity phases.

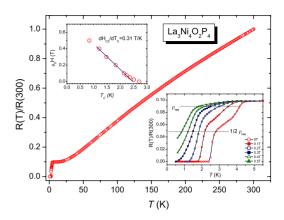


FIG. 5. (Color online) Temperature dependence of electrical resistance R(T)/R(300 K). The upper inset shows the temperature dependence of the upper critical field (H_{C2}) from the midpoint of the superconducting transition $[\rho(T_c)=1/2\rho_{\rm res.}]$. Lower inset shows resistance data at low temperature, illustrating the changes in the superconducting transition with applied magnetic field. The presence of the LaNiPO minority phase results in a partial transition starting near 4.5 K that is suppressed quickly in the applied field.

culated from the $\chi(T)$ fit of 50–350 K to Curie-Weiss law]. This is significantly higher than the free-electron value (R_W = 1) and indicates a spin-enhanced susceptibility. The normalized specific-heat jump at T_c can also be calculated from γ . The experimental value for La₃Ni₄P₄O₂, $\Delta C_{\rm el.}/\gamma T_c$ =1.25, is lower than the BCS prediction²⁰ of 1.43, likely due to the influence of the impurity phases on the measured γ . The simple Debye model connects the β coefficient and Debye temperature (Θ_D) through $\Theta_D = (\frac{12\pi^4}{5\beta_m}nR)^{1/3} \approx 360$ K, where R=8.314 J mol⁻¹ K⁻¹, β_m =4 β =0.56 mJ mol K⁻⁴, and n=13 for La₃Ni₄P₄O₂. With this value for Θ_D , we can estimate the electron-phonon coupling constant ($\lambda_{\rm ep}$) from McMillian's relation:²¹ $\lambda_{\rm ep} = \frac{1.04 + \mu^* \ln(\Theta_D/1.45T_c)}{(1-0.62\mu^*)\ln(\Theta_D/1.45T_c)-1.04}$, where μ^* is a Coulomb repulsion constant. Taking a typical value for μ^* =0.10, we estimate $\lambda_{\rm ep} \approx 0.5$, which implies that La₃Ni₄P₄O₂ is a weakly to moderately coupled superconductor, consistent with the specific-heat jump at T_c .

Electrical resistivity measurements are shown in the main panel of Fig. 5. Between room temperature and 5 K, the resistivity decreases by nearly a factor of 10, which is reasonable for a polycrystalline sample. The lower inset shows the resistivity near T_c for representative applied fields. The higher transition is associated with the LaNiPO impurity, which superconducts near 4 K. Due to the presence of this superconducting impurity phase, we use the temperature at which the resistivity reaches half its normal-state value, $\rho(T)=1/2$ ρ_{res} , to define the midpoint T_c of La₃Ni₄P₄O₂. With this definition of T_c , the upper inset of Fig. 5 presents the temperature dependence of the estimated upper critical field (H_{C2}) . The initial slope $-dH_{C2}/dT_c = 0.31(1)$ T/K permits an estimate the zero-temperature upper critical field $H_{C2}(0)$ from $0.7T_c dH_{C2}/dT_c \approx 6$ kOe.²² With this estimate for $H_{C2}(0)$, the zero-temperature superconducting coherence length can be estimated¹⁸ as $\xi = [\phi_0/2\pi H_{C2}(0)]^{1/2}$ $\approx 240 \pm 20$ Å, where ϕ_0 is the flux quantum. This is a reasonable value, but is significantly larger than that found in



the related Ni-based superconductors YNi₂B₂C (65 Å) (Ref. 23) and MgCNi₃ (47 Å).²⁴ The present analysis does not reveal any overtly unconventional behavior in La₃Ni₄P₄O₂ beyond the presence of a spin-enhanced susceptibility. While there is widespread belief that there is a relationship between superconductivity and magnetism in the Fe-based oxypnictides, early band-structure calculations on LaNiPO (Ref. 25) have been interpreted as indicating that the nickel oxypnictides are conventional superconductors. The issue of conventional vs unconventional superconductivity in the Ni-based superconductors—the layered borocarbides MgCNi₃—continues to be an open one, however. Arguments both for and against conventional superconductivity in this family, based on detailed experiments, continue to be presented. 19,25-28 Thus, future detailed characterization of the nickel-based oxypnictide superconductors La₃Ni₄P₄O₂, and LaNiPO will be of interest. The crystal structure of La₃Ni₄P₄O₂ is unique among the superconducting oxypnictides due to the asymmetric distribution of charge reservoir layers around the $\rm Ni_2P_2$ plane. For the one copper oxide superconductor in which a similar structural asymmetry is known, $\rm Y_2Ba_4Cu_7O_{15}$, unexpected behavior has been predicted.²⁹ Whether this asymmetry may be the reason why the T_c in $\rm La_3Ni_4P_4O_2$ is approximately half that of $\rm BaNi_2P_2$ and $\rm LaNiPO$ is an issue that will require further study. The growth of single crystals will be required to provide more precise characterization of the normal state and superconducting properties.

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- ¹J. G. Bednorz and K. A. Z. Müller, Z. Phys. B: Condens. Matter 64, 189 (1986).
- ² Y. Kamihara, H. Hiramatsu, M. Hirano, R. Kawamura, H. Yanagi, T. Kamiya, and H. Hosono, J. Am. Chem. Soc. 128, 10012 (2006).
- ³T. Watanabe, H. Yanagi, T. Kamiya, Y. Kamihara, H. Hiramatsu, M. Hirano, and H. Hosono, Inorg. Chem. **46**, 7719 (2007).
- ⁴Y. Kamihara, Y. Kamihara, T. Watanabe, M. Hirano, and H. Hosono, J. Am. Chem. Soc. **130**, 3296 (2008).
- ⁵F. C. Hsu, J. Y. Luo, K. W. Yeh, T. K. Chen, T. W. Huang, P. M. Wu, Y. C. Lee, Y. L. Huang, Y. Y. Chu, D. C. Yan, and M. K. Wu, Proc. Natl. Acad. Sci. U.S.A. **105**, 14262 (2008).
- ⁶X. C. Wang, Q. Q. Liu, Y. X. Lv, W. B. Gao, L. X. Yang, R. C. Yu, F. Y. Li, and C. Q. Jin, Solid State Commun. **148**, 538 (2008).
- ⁷M. J. Pitcher, D. R. Parker, P. Adamson, S. J. C. Herkelrath, A. T. Boothroyd, R. M. Ibberson, M. Brunelli, and S. J. Clarke, Chem. Commun. (Cambridge) (2008) 5918.
- ⁸T. Mine, H. Yanagi, T. Kamiya, Y. Kamihara, M. Hirano, and H. Hosono, Solid State Commun. **147**, 111 (2008).
- ⁹M. Rotter, M. Tegel, and D. Johrendt, Phys. Rev. Lett. **101**, 107006 (2008).
- $^{10}\rm Ni_5P_4$ was synthesized from Ni powder (Alfa, 99.5%) and red P (Alfa 99.9%) by slow heating to 800 °C. La shavings (Johnson Matthey, 99.9%), Ni₅P₄, P, and dried La₂O₃ (Alfa 99.9%) were mixed, pelletized, placed in an alumina crucible in an evacuated quartz tube, and heated from 750 °C to 1050 °C over 1 hr. The sample was then reground, repressed, and reheated, with 2% excess P, at 1200 °C overnight. Finally, the sample was reground, repressed, and reheated at 1200 °C overnight (three times) in a tube backfilled with 1/3 atm Ar.
- ¹¹Diffraction experiments were performed with neutrons of wavelength 1.5403 Å produced by a Cu(311) monochromator. Collimators with horizontal divergences of 15′, 20′, and 7′ of arc were used before and after the monochromator, and after the sample, respectively. Data were collected in the 2θ range of 3° – 168° with a step size of 0.05° . The neutron-scattering am-

- plitudes used in the refinements were La=0.824, Ni=1.030, P =0.513, and O=0.581 \times 10⁻¹² cm.
- ¹²B. H. Toby, J. Appl. Crystallogr. **34**, 210 (2001).
- ¹³A. C. Larson and R. B. Von Dreele, Los Alamos National Laboratory LAUR, Report No. 86, 2000 (unpublished).
- ¹⁴J. W. Kaiser and W. Jeitschko, Z. Naturforsch. B 57, 165 (2002).
- ¹⁵V. Vildosola, L. Pourovskii, R. Arita, S. Biermann, and A. Georges, Phys. Rev. B 78, 064518 (2008).
- ¹⁶R. Hoffmann and C. Zheng, J. Phys. Chem. **89**, 4175 (1985).
- ¹⁷W. K. Hofmann and W. Jeitschko, J. Solid State Chem. **51**, 152 (1984).
- ¹⁸Due to the polycrystalline nature of the samples, this quantity is averaged over all crystallographic directions.
- ¹⁹ M. Nohara, M. Isshiki, H. Takagi, and R. J. Cava, J. Phys. Soc. Jpn. **66**, 1888 (1997).
- ²⁰J. Bardeen, L. N. Cooper, and J. R. Schrieffer, Phys. Rev. **108**, 1175 (1957).
- ²¹W. L. McMillan, Phys. Rev. **167**, 331 (1968).
- ²²N. R. Werthamer, E. Helfand, and P. C. Hohenberg, Phys. Rev. 147, 295 (1966).
- ²³ H. Takagi, H. M. Nohara, and R. J. Cava, Physica B **237-238**, 292 (1997).
- ²⁴Z. Q. Mao, M. M. Rosario, K. D. Nelson, K. Wu, I. G. Deac, P. Schiffer, Y. Liu, T. He, K. A. Regan, and R. J. Cava, Phys. Rev. B 67, 094502 (2003).
- ²⁵A. Subedi, D. J. Singh, and M. H. Du, Phys. Rev. B 78, 060506(R) (2008).
- ²⁶H. S. Lee, D. J. Jang, H. G. Lee, S. I. Lee, S. M. Choi, and C. J. Kim, Adv. Mater. **19**, 1807 (2007).
- ²⁷ T. Watanabe, M. Nohara, T. Hanaguri, and H. Takagi, Phys. Rev. Lett. **92**, 147002 (2004).
- ²⁸B. Bergk, V. Petzold, H. Rosner, S. L. Drechsler, M. Bartkowiak, O. Ignatchik, A. D. Bianchi, I. Sheikin, P. C. Canfield, and J. Wosnitza, Phys. Rev. Lett. **100**, 257004 (2008).
- ²⁹S. A. Kivelson, Physica B 318, 61 (2002); S. A. Kivelson (private communication).

